



CALIFORNIA STATE SCIENCE FAIR
2015 PROJECT SUMMARY

Name(s) Hope Lee	Project Number S1016
Project Title A Novel Method to Immobilize Ionic Liquid in Alginate-Gelatin Polymer Beads for Heavy Metal(s) Removal	
<div>Objectives/Goals<p>The purpose of this scientific study was to design a novel, effective, and highly efficient immobilized ionic liquid approach towards removal of heavy metal(s) from an aqueous environment. It was predicted that increased amounts of ionic liquid would remove a greater proportion of the Cu^{2+} from the aqueous environment and that the system would be superior to traditional LLE in terms of efficiency, economic feasibility, and environmental impact.</p></div> <div>Abstract<p>The ionic liquid chosen was trihexyl (tetradecyl) phosphonium bis (2,4,4-trimethylpentyl) phosphinate (CYPHOS IL 104), a synthesized organic compound which consists of ions of both charges and is liquid at room temperature. For this study, copper ion (Cu^{2+}) was selected as a model system to demonstrate the immobilized CYPHOS IL 104 concept. Gelatin and sodium alginate were selected to immobilize and stabilize IL in a polymer matrix.</p></div> <div>Methods/Materials<p>The ionic liquid chosen was trihexyl (tetradecyl) phosphonium bis (2,4,4-trimethylpentyl) phosphinate (CYPHOS IL 104), a synthesized organic compound which consists of ions of both charges and is liquid at room temperature. For this study, copper ion (Cu^{2+}) was selected as a model system to demonstrate the immobilized CYPHOS IL 104 concept. Gelatin and sodium alginate were selected to immobilize and stabilize IL in a polymer matrix.</p></div> <div>Results<p>The optimized composition for the immobilized ionic liquid solution was identified through a DOE model as approximately 0.33% gelatin (w/w), 0.33% sodium alginate (w/w), and 33% IL (w/w). The immobilized IL beads ultimately removed a maximum of over 98% of Cu^{2+} from 6 mL of 50 mM Cu^{2+} solution. No extraction efficiency was compromised through immobilization. The immobilized IL beads were successfully stripped and regenerated by 1N NaOH and was able to remove over 98% of Cu^{2+} from 6 mL of 50 mM Cu^{2+} solution when used again.</p></div> <div>Conclusions/Discussion<p>There are several large benefits to this immobilized ionic liquid approach: [1] drastic reduction in overall processing time, [2] reduction of CYPHOS# IL 104 material lost during process, [3] reduction of the chance of CYPHOS# IL 104 contamination in the water, [4] units more portable and easier to handle (can be handled dry as well), and [5] significant reduction of the total cost from raw materials and operation. This developed approach abides by many Green Chemistry principles.</p></div>	
Summary Statement <p>The purpose of this scientific study was to design a novel, effective, and highly efficient immobilized ionic liquid approach towards removal of heavy metal(s) from an aqueous environment.</p>	
Help Received <p>Father supervised while working at home</p>	