



# CALIFORNIA SCIENCE & ENGINEERING FAIR 2019 PROJECT SUMMARY

<b>Name(s)</b>  <b>Alexis MacAvoy</b>	<b>Project Number</b>  <b>J0619</b>
<b>Project Title</b>  <b>Designing Efficient, Low-Cost, Eco-Friendly Activated Carbon for Removal of Heavy Metals from Water</b>	
<p style="text-align: center;"><b>Abstract</b></p> <p><b>Objectives</b> Heavy metal contamination continues to threaten the SF Bay &amp; cleanup of mercury &amp; copper from watersheds is costly &amp; time-consuming. Designing a cheap &amp; effective activated carbon (AC) for filtration of industrial waste is an unmet need for heavy metal adsorption from water.</p> <p><b>Methods</b> Biowaste materials (coconut &amp; walnut shells, sawdust, ash) were used as precursors because of high lignin &amp; cellulose content. Precursors were treated with dilute sulfuric acid to increase lignin &amp; cellulose content. Adsorbents were activated using 85% phosphoric acid (288 C x 6h). For Phase 2, repeated 8% NaHCO<sub>3</sub> rinses were performed to raise pH. Adsorbents were treated with 15% NaF or NaCl halogen solution to increase mercury (Hg) s affinity to the carbons. These activated carbons were designed for Hg adsorption but given Hg s toxicity, copper (Cu) was used as a test contaminant. All ACs were packed in burettes &amp; 10mL 0.02 g Cu/L solution introduced as a contaminant. To test effluents from AC columns, 3 quantitative spectrophotometric assays &amp; 1 qualitative assay were performed.</p> <p><b>Results</b> 3 spectrophotometric assays were attempted to determine Cu concentration in effluents &amp; the Cu-ammonium-cuprizone assay was most accurate &amp; reproducible. During Phase 1, the most efficient ACs were the sawdust &amp; coconut controls. Cu levels in effluents were reduced to levels of Cu remaining in commercially produced AC, but control carbons achieved similar results to treated carbons. In Phase 2 synthesis, the objective was to rinse the carbons to neutralize pH. Testing of effluents from Phase 2 chemically treated carbons demonstrated a marked drop in Cu concentrations compared to Phase 1 syntheses, concluding that high pH increased Cu removal.</p> <p><b>Conclusions</b> Low-cost, AC syntheses removed the same amount of Cu as the commercially available AC for the Phase 1 synthesis, &amp; up to 30 times more for the Phase 2. About 90% of the test Cu(II) on average was removed on the first pass through the test carbon column in Phase 1 &amp; up to 99.67% of test Cu(II) was removed in Phase 2 to levels below the allowable EPA Cu concentration for drinking water. Home design &amp; synthesis of effective AC using low-cost materials is a usable option for heavy metal adsorption, especially in high-risk environments with few resources.</p>	
<b>Summary Statement</b>  I synthesized and tested cheaper, effective, eco-friendly activated carbon that removed 99.67% of copper from a standard copper solution.	
<b>Help Received</b>  My parents helped me wash dishes and let me use the garage to conduct experiments.	